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rotational

equilibrium: Ex.

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$$(16/18) * 100 =$$

88.9% Percentage
by mass of
element in a compound

= (mass
of element in 1
mol of compound
÷ molar mass of

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compound) $\times 100$

% (after
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#1) Find the %

of carbon in

sodium

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(NaHCO_3).

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Percentage by

mass of element

in a compound =

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